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(a) Field of the Invention This invention generally relates to agricultural compositions, and more particularly to high nitrogen content liquid binder compositions for use in the storage and handling of aerial tubers. (b) Description of the Prior Art Most commercial operations involving growing tubers utilize certain agricultural chemicals such as fertilizer, herbicides, pesticides, etc. These chemicals are mixed together and applied to the leaves of the plant. Some of the chemicals may be absorbed by the leaves and other parts of the plant, and thus are taken into the tubers as they develop. These chemicals, if present in the tubers as they emerge, may adversely affect the quality of the tubers. Thus, it is generally desirable to apply the chemicals to the leaves only and not directly to the tubers. In commercial operations involving potato or onion growing, the tubers are brought to the ground and handled in storage tanks. Some of the chemicals presently used are not readily miscible or soluble in the water normally used in the storage tanks of a commercial operation. This is especially true when the storage tanks are used to store the tubers for relatively long periods of time. In commercial potato operations, it is common to use a binder to attach the fertilizer to the leaves of the plants rather than to the tubers. The typical binder for this purpose is ammonium nitrate. This fertilizer, however, is relatively expensive and thus increases the cost of the potatoes. Various synthetic binders have been used in order to overcome the problems discussed above. One such binder is an aqueous solution of a mixture of urea and urea formaldehyde. While this binder is relatively inexpensive, it has a number of drawbacks. For example, the binder has an undesirable odor. It also has a relatively short shelf life. In some instances, it is not possible to add fertilizer to the solution as the solution ages. The solution has a low viscosity at high temperatures, and thus will not adhere to the leaves well at temperatures above 60 degrees F. Another binders known in the art are fluorescein and polyvinyl acetate binder. The fluorescein binders have the disadvantage that they have a very short shelf life. The shelf life is only one year when stored at 60 degrees F. or below. The shelf life of the polyvinyl acetate binder is even shorter. A binder which is economical and desirable from several standpoints would be an aqueous solution of urea

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